

**Special Case Double Displacement Reactions Practice**

Answer the following question on a separate piece of paper. Determine the products that form in the double displacement reaction between each pair of reactants. Write the balanced chemical equation for the reaction.

1. sodium hydroxide and hydrobromic acid
2. potassium carbonate and hydrochloric acid
3. ammonium chloride and magnesium hydroxide
4. sulphuric acid and sodium carbonate
5. calcium carbonate and acetic acid
6. rubidium hydroxide and hydrochloric acid
7. lithium hydroxide and ammonium bromide
8. sulfuric acid and lithium hydroxide
9. ammonium bromide and sodium hydroxide
10. calcium hydroxide and nitric acid